

8th International Food Safety Congress

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Grand Cevahir Hotel & Convention Center



Emerging Food Safety Risks **Thermal Process Contaminants**

Vural Gökmen



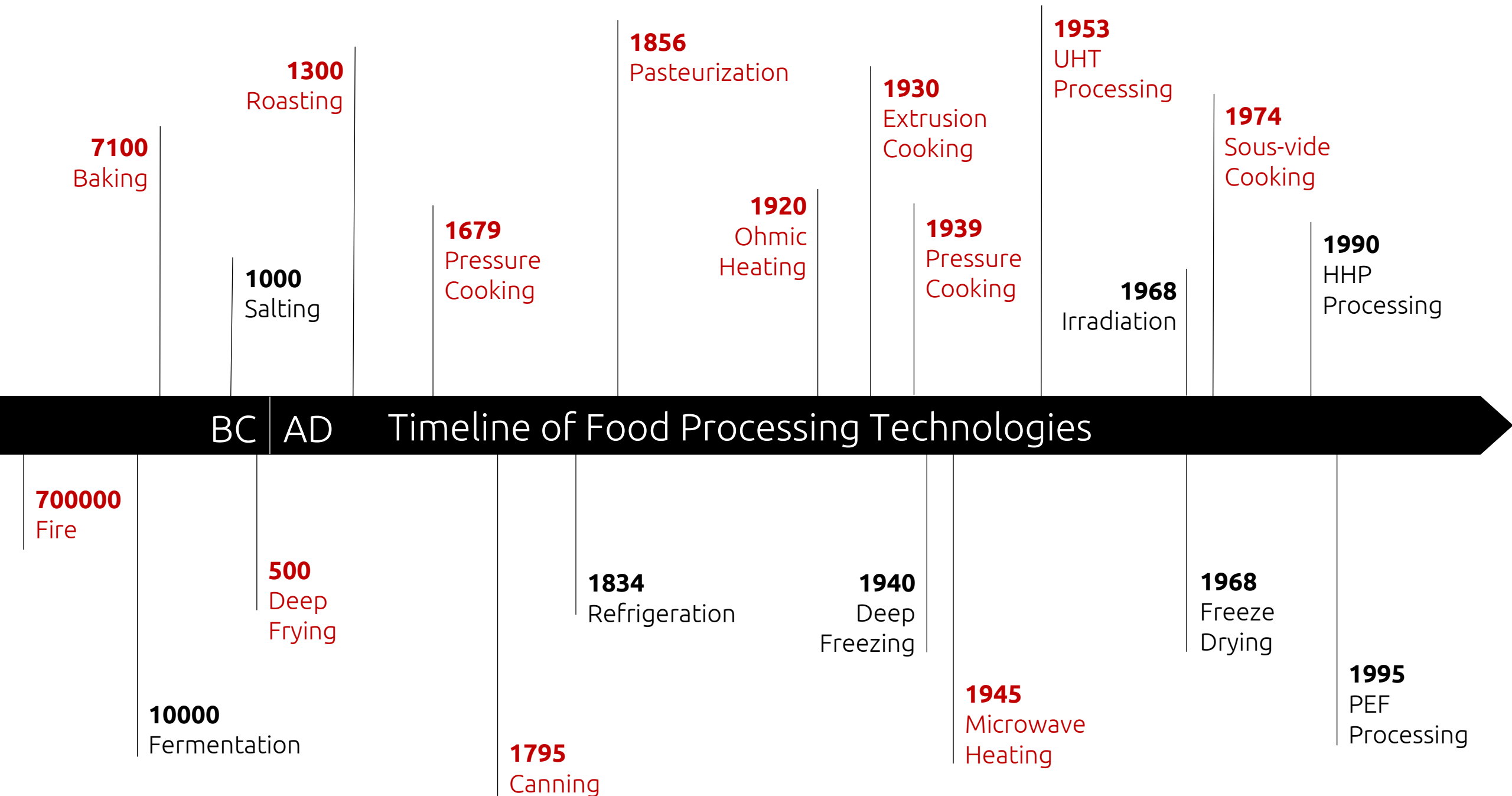
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Thermally Processed Foods



Baking
150-250°C

Roasting
150-250°C

Frying
170-190°C

Pasteurization
70-100°C

Sterilization
110-130°C

Pasteurization
70-100°C

Pasteurization
70-100°C
Sterilization
110-130°C

Deodorization
150-260°C

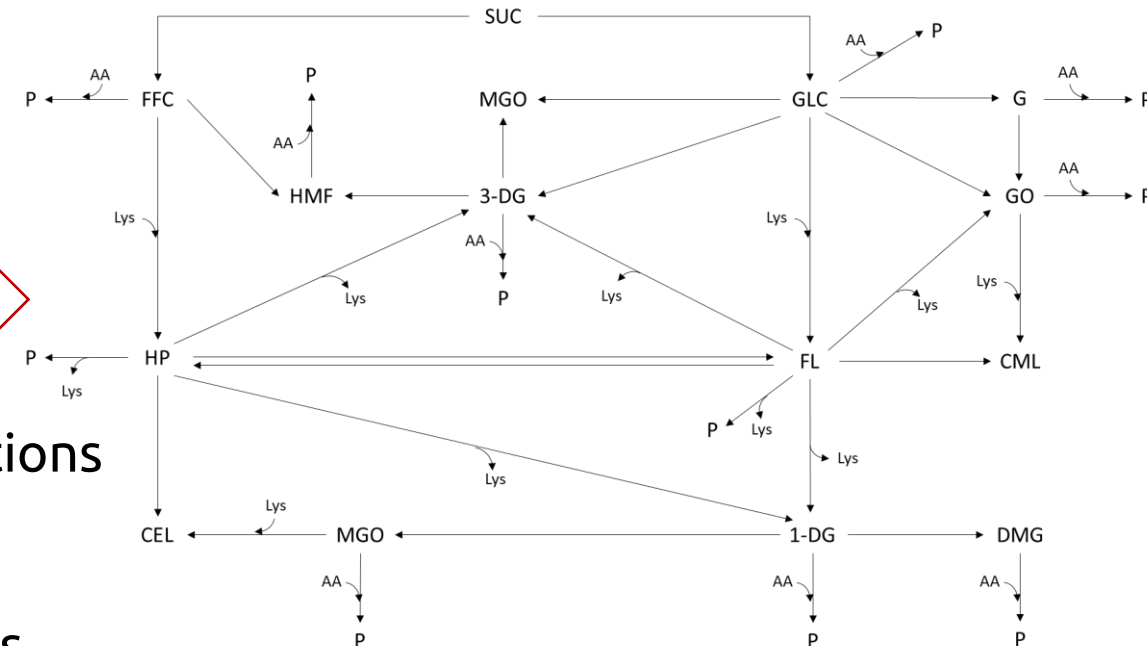
Roasting
150-200°C

Food Composition (Reaction Mixture)

- Carbohydrates
- Proteins
- Lipids
- Minerals
- Vitamins
- Water

Heating / Cooking

- T/t induced complex chemical reactions
- Changes in food composition
- Effects on food quality and safety
- Considerations on risks and benefits
- Need for product reformulations and/or process modifications



Thermal Process Contaminants

- Precursors Naturally Found in Foods**
- Asparagine
 - Ascorbic acid
 - Sugars
 - Polyunsaturated fatty acids
 - Acylglycerols
 - Creatin(in)e

Thermally Processed Food Products



Bakery Products

Coffee Products

Potato Products

Honey Products

Baby Foods

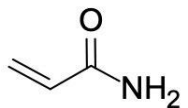
Juice Products

Milk Products

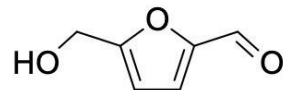
Vegetable Oils

Meat Products

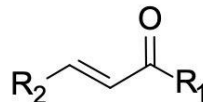
Acrylamide



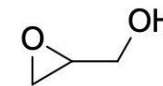
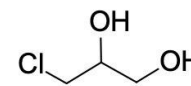
Furan Derivatives



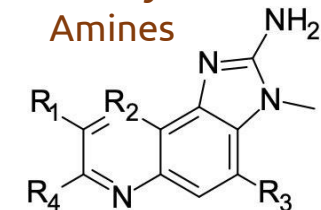
Dicarbonyl Compounds



Chloropropanols and Glycidyl Esters



Heterocyclic Amines



Acrylamide

Maximum limits are not regulated in EU.

French fries

Mean 338 $\mu\text{g/kg}$
Max 2174 $\mu\text{g/kg}$

Biscuits, crackers, crisp bread

Mean 680 $\mu\text{g/kg}$
Max 3307 $\mu\text{g/kg}$

Soft bread

Mean 53 $\mu\text{g/kg}$
Max 563 $\mu\text{g/kg}$

Potato crisps

Mean 675 $\mu\text{g/kg}$
Max 4533 $\mu\text{g/kg}$

Breakfast cereals

Mean 155 $\mu\text{g/kg}$
Max 2072 $\mu\text{g/kg}$

Processed cereal based foods for infants and young children

Mean 69 $\mu\text{g/kg}$
Max 1200 $\mu\text{g/kg}$

Coffee and coffee substitutes

Mean 393
 $\mu\text{g/kg}$
Max 7095
 $\mu\text{g/kg}$

The Benchmark Levels



Instant coffee
850 µg/kg

Roast Coffee
400 µg/kg



Biscuits and Crackers
350 µg/kg



Soft Bread
50 µg/kg



Potato Crisps
750 µg/kg



French Fries
500 µg/kg



Baby Foods
40 µg/kg

Acrylamide

- The **Maillard reaction** is responsible for the formation of acrylamide in foods during heating at elevated temperatures.
- **Asparagine** which is a **major precursor** of acrylamide occurs naturally in plants.
- Asparagine concentration is considered as **the limiting factor** for acrylamide formed in foods. In practice, the foods rich in free asparagine and reducing sugars have the highest potential to form acrylamide.
- **Carbonyl compounds** are known to **accelerate** acrylamide formation during heating. Foods may be considered as a pool of carbonyl compounds.

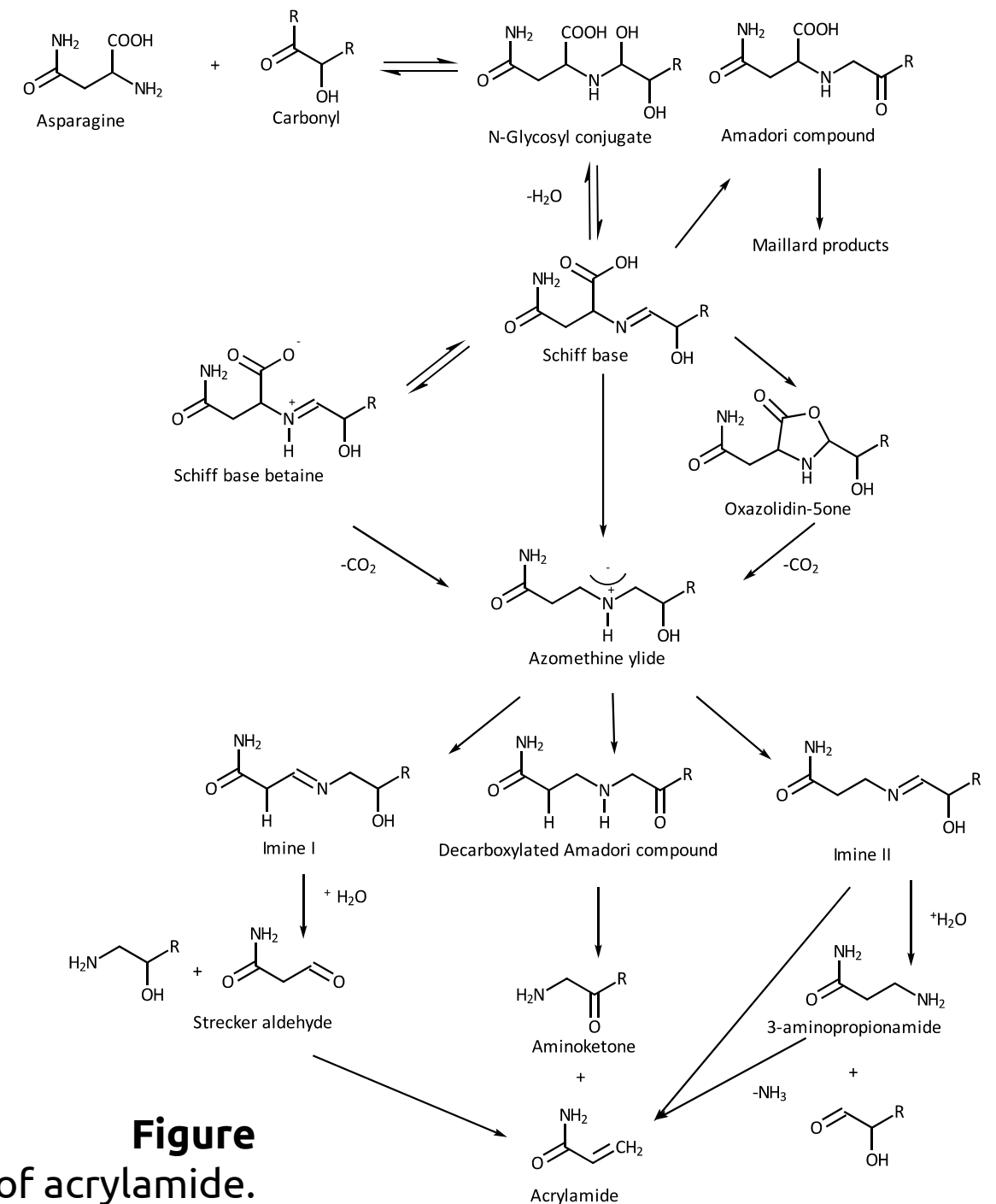


Figure
Formation mechanism of acrylamide.

Mechanistic Basis of Mitigation



Targeting asparagine as the limiting factor

- Decreasing asparagine level as low as possible by selecting cereals with low asparagine, diluting with low asparagine cereals, using asparaginase, extending fermentation time where possible
- Blocking carboxyl group of asparagine by adding divalent cations

Targeting carbonyl/dicarbonyl compounds as the accelerating factor

- Avoiding or minimizing reducing sugars, replacing fructose with glucose, using glucose oxidase, extending fermentation time where possible
- Limiting sugar fragmentation reactions leading to α -dicarbonyl and α -hydroxy carbonyl compounds by replacing ammonium bicarbonate with alternatives where feasible
- Trapping sugar fragmentation products by adding thiol and/or phenolic compounds
- Addition of amino acids to compete with asparagine for carbonyl compounds

Targeting thermal load and T/t history of the baking process

- Decreasing overall thermal load by adjusting T/t during baking
- Decreasing surface-to-volume ratio of product
- Avoiding very high baking temperatures and maintenance of uniform control of the color

Furan

Maximum limits are not regulated in EU.

**Boiling point
31.3°C**

Roasted Coffee & Jarred Baby Purees

Furan is a volatile compound formed mostly during the thermal processing of foods. The toxicity of furan has been well documented previously, and it was classified as “possible human carcinogen (Group 2B)” by the International Agency for Research on Cancer.



Furan

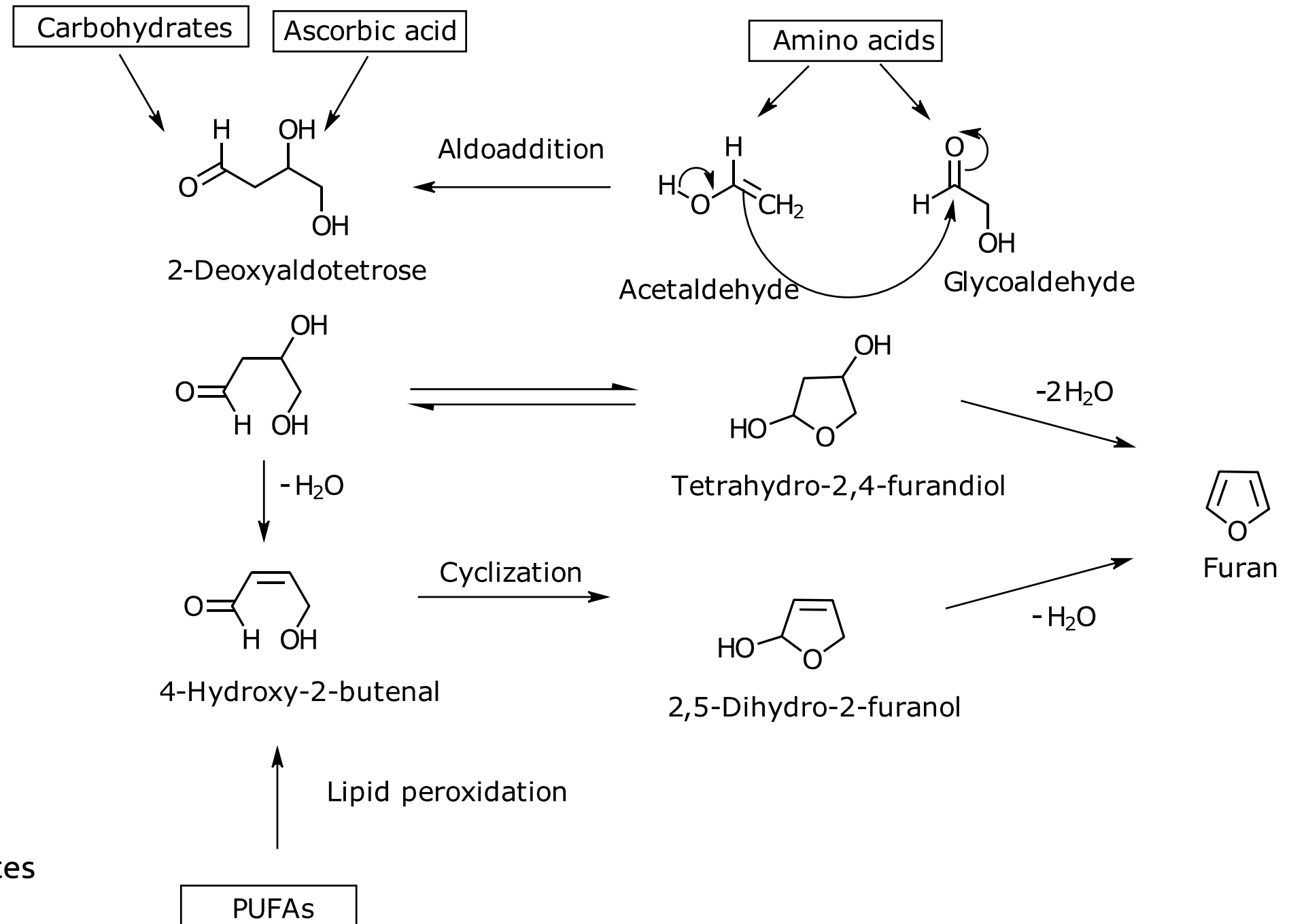


Figure
Summary of possible routes
for furan formation

Chloropropanols

A close-up photograph of a glass bottle with a yellow cap, tilted to pour a golden-yellow liquid (oil) into a black frying pan. The background is blurred, showing green foliage.

Maximum limits are regulated in EU.

3-MCPD and its esters

- 20 µg/kg in hydrolyzed vegetable protein and soy sauce
- 1250 – 2500 µg/kg in vegetable oils and fats
- 15 µg/kg in infant formula

Glycidyl esters

- 500 – 1000 µg/kg in vegetable oils and fats
- 6 µg/kg in infant formula

Fats and Oils

Chloropropanols and related substances are contaminants that can be formed during food processing, particularly when high temperatures are applied to fats and oils. The original maximum levels for 3-MCPD and its esters in specific food products for infants and young children were established by Regulation (EU) 2020/1322.

Chloropropanols

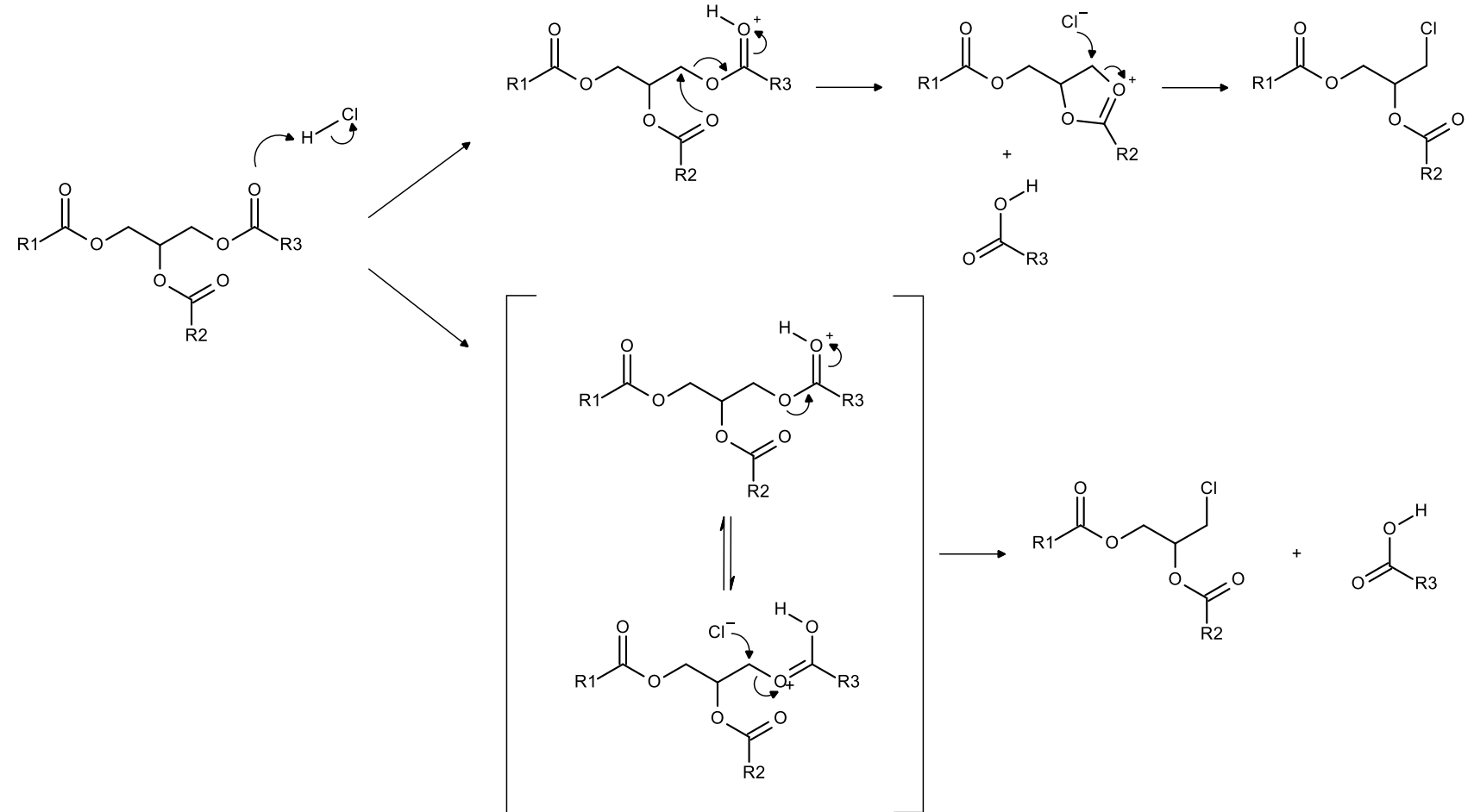
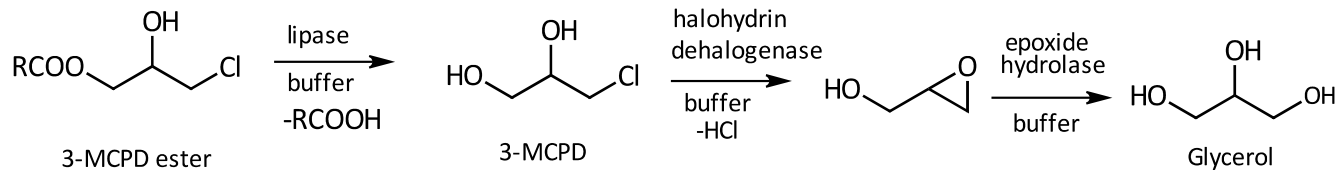
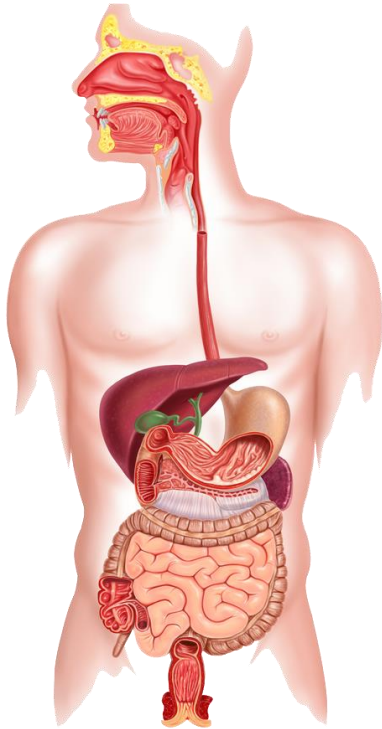
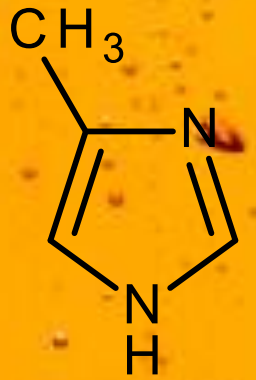


Figure
Mechanism of the formation of chloropropanols and related substances from acrylglycerols

Methylimidazole



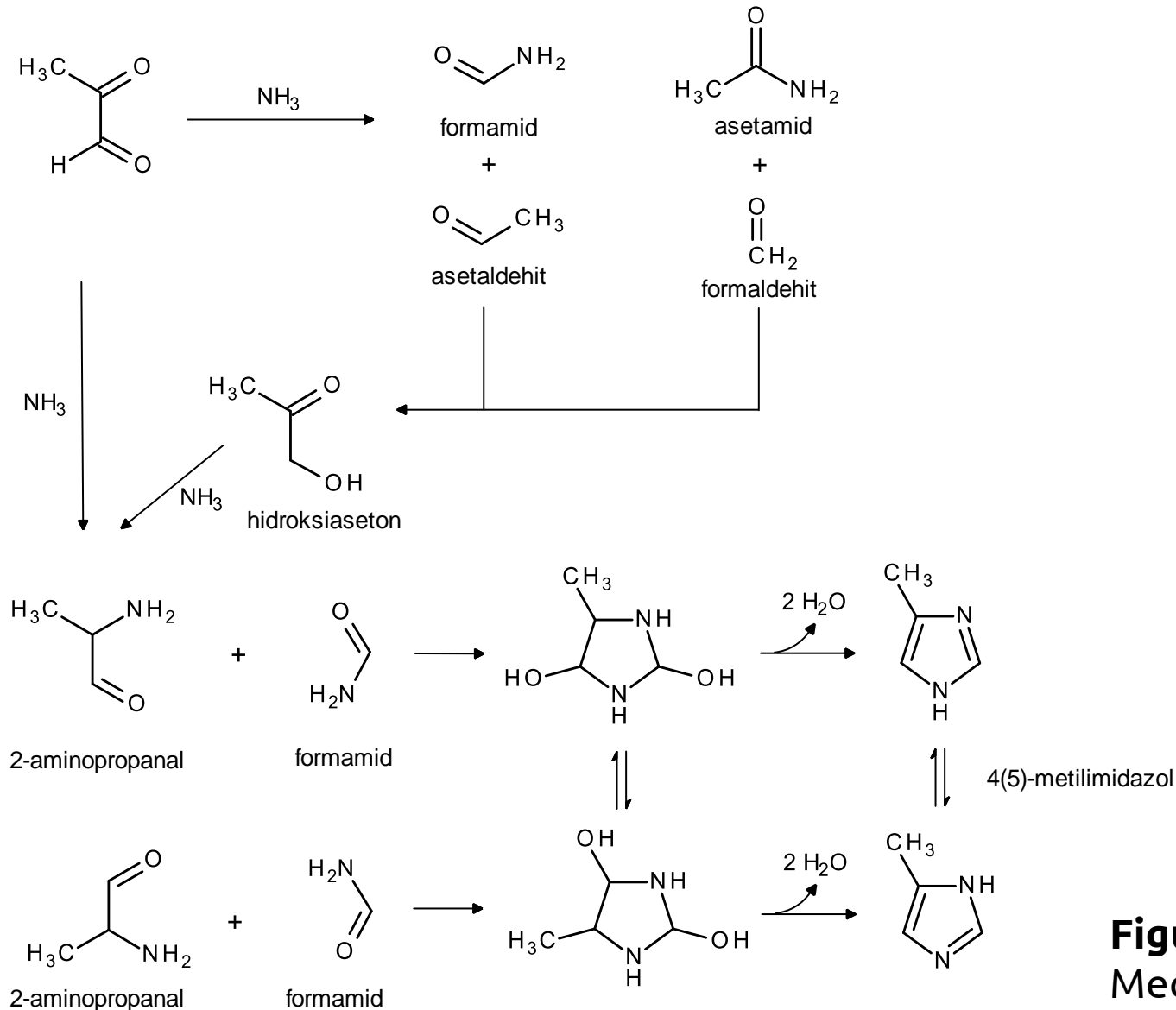
Maximum limits are not regulated in EU.

- Caramel Color I (also known as plain or spirit caramel)
- Caramel Color II (caustic sulfite caramel)
- **Caramel Color III** (ammonia or beer caramel, baker's and confectioner's caramel)
- **Caramel Color IV** (known as sulfite-ammonia, soft drink caramel or acid-proof caramel)

Coke, Beer, Candy, Bakery Products

The International Agency for Research on Cancer classified methylimidazole (-4-MEI) as Group 2B indicating sufficient evidence of carcinogenicity in experimental animals while no human data were available. In 2011, California listed 4-MEI as a carcinogen under the Proposition 65.

Methylimidazole



Ammonia Caramel



Figure
Mechanism of the formation of methylimidazole

AGEs in Foods

A close-up photograph of a blue milk carton pouring white milk into a clear glass filled with ice cubes. The milk is captured mid-pour, creating a dynamic splash and bubbles. The background is slightly blurred, showing more of the carton and a hint of a green plant on the right.

Maximum limits are not regulated in EU.

- In vitro
- In vivo

UHT-Treated and Powdered Milk Products

Advanced glycation end products (AGEs) are proteins that become glycated as a result of exposure to sugars.

They are a biomarker implicated in aging and the development or worsening of many degenerative diseases, such as diabetes, atherosclerosis, chronic kidney disease, and Alzheimer's disease.

Maillard Reaction in Foods

$T > 100^{\circ}\text{C}$

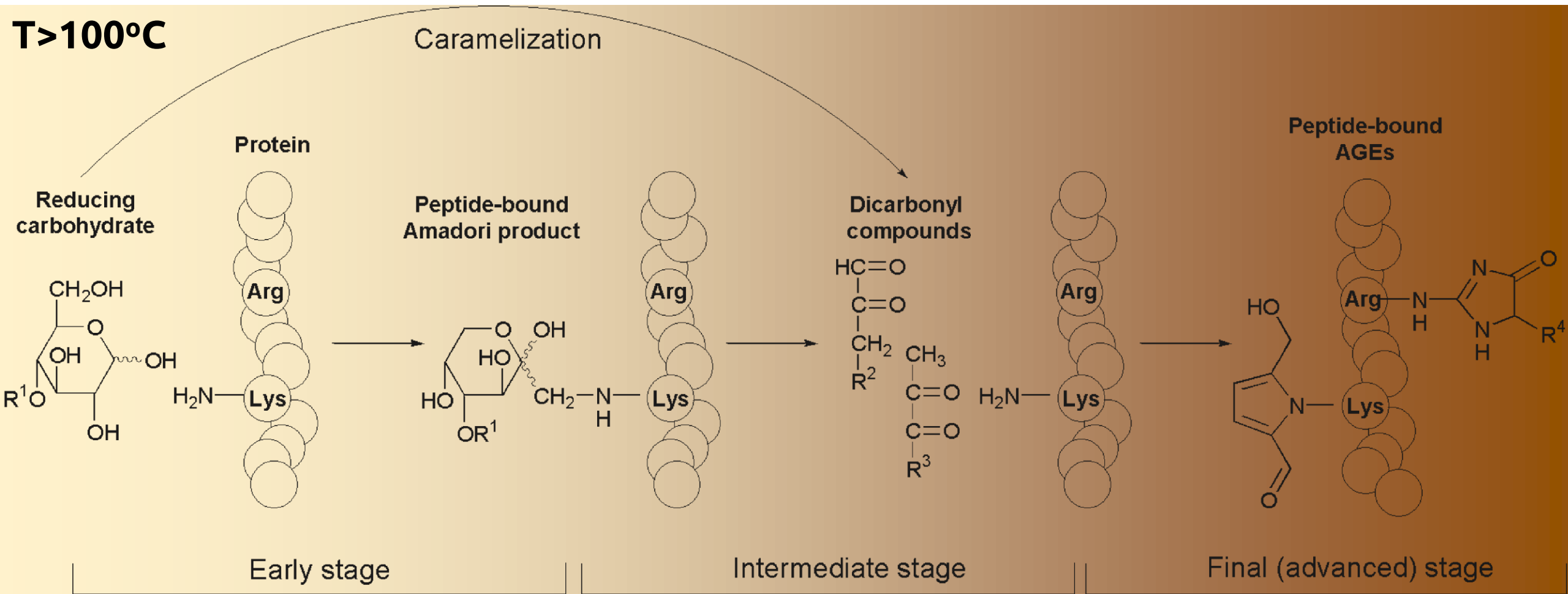


Figure
General reaction scheme of the Maillard reaction leading to the formation of dicarbonyl compounds and advanced glycation end products (AGEs) in foods during thermal processing.

Maillard Reaction in Human Body

T=37°C

food

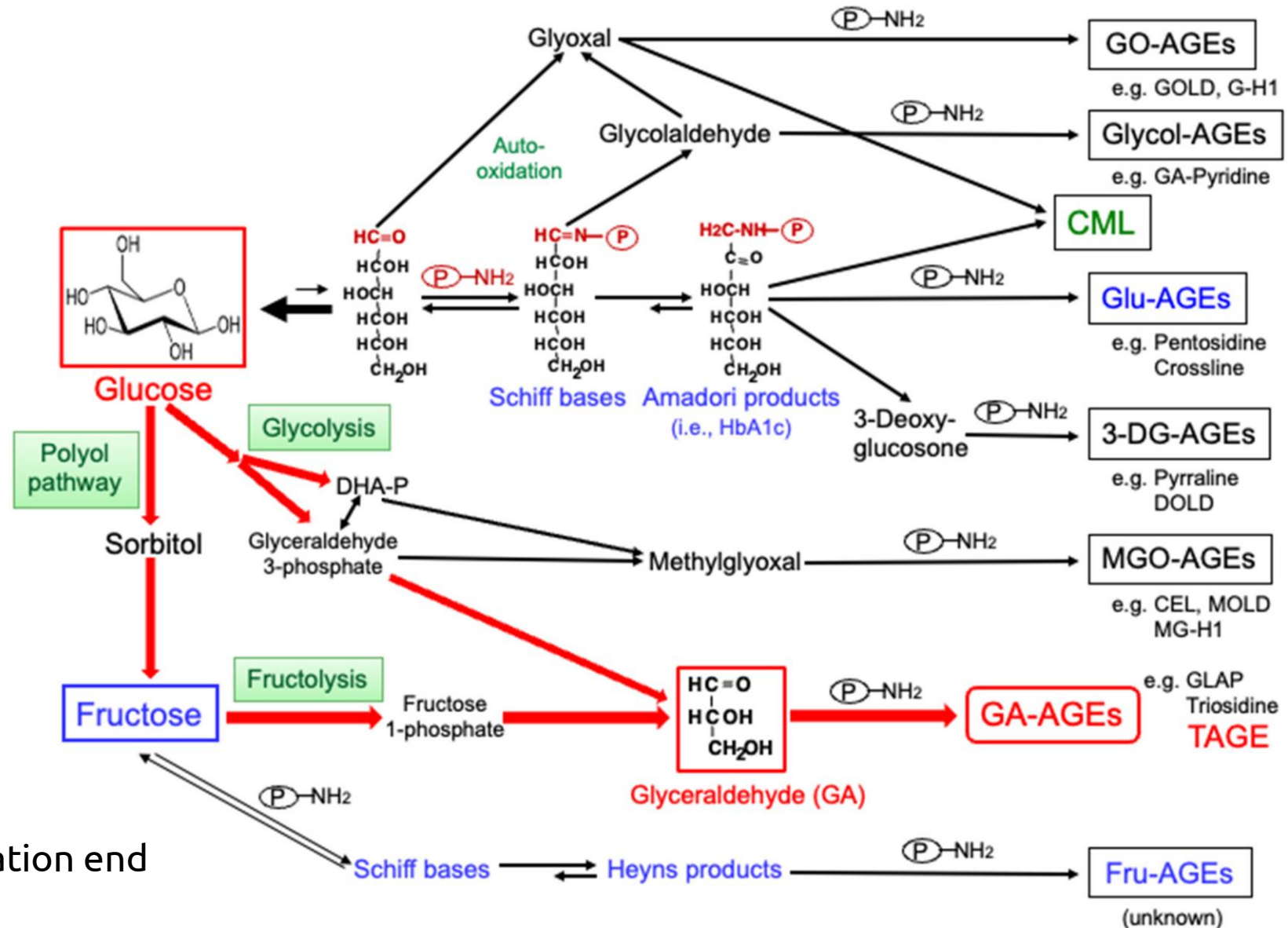
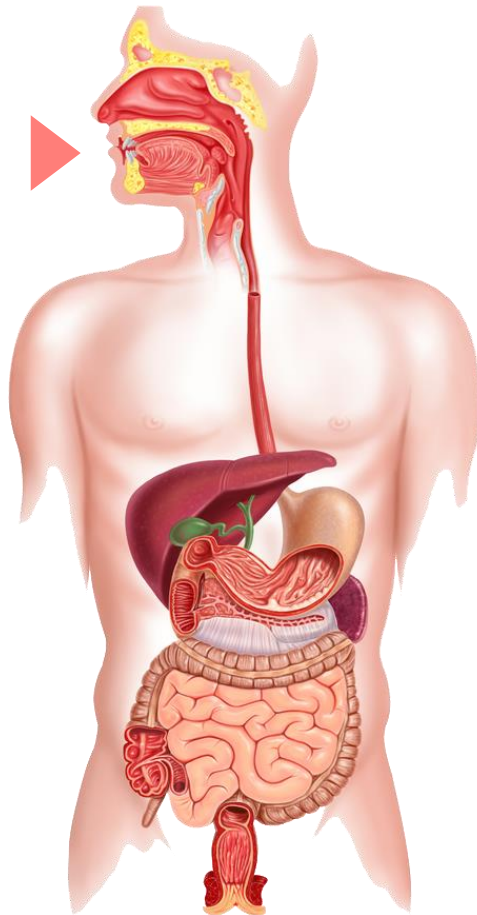
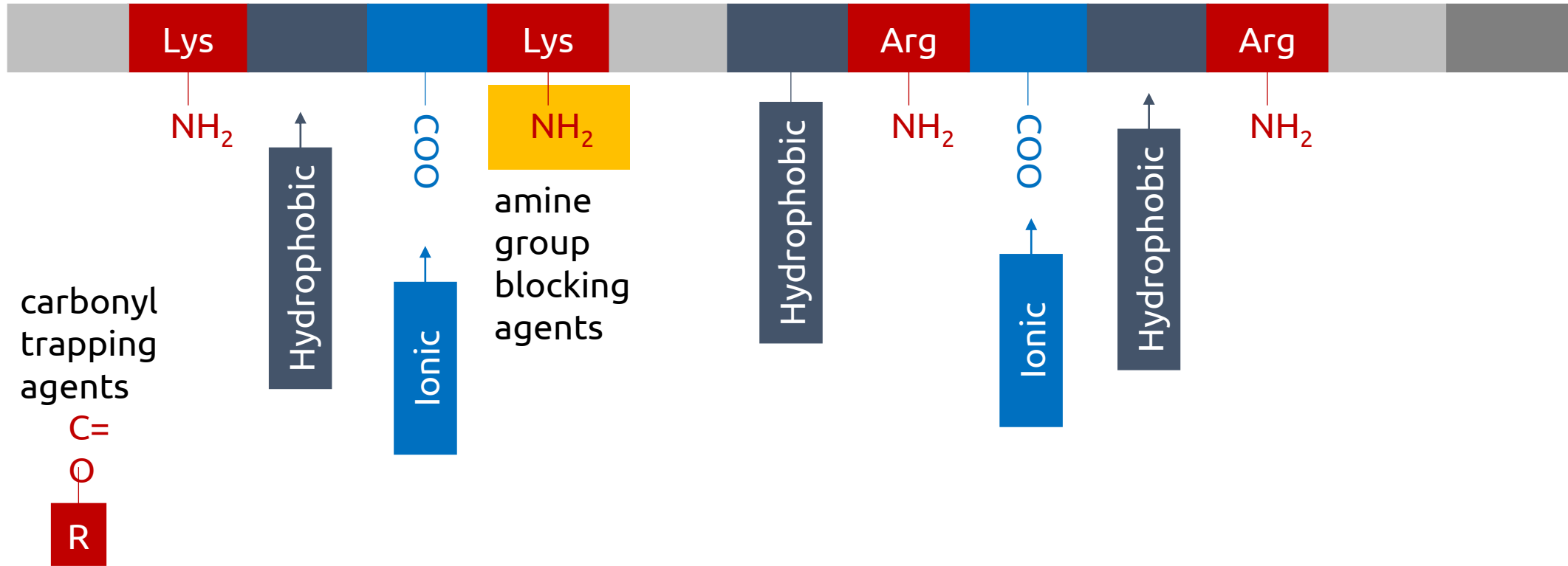


Figure
Formation mechanism of advanced glycation end products (AGEs) in the human body.

Anti-Glycation Scenarios



steric hindrance of free amino groups by hydrophobic (polyphenols) or ionic interactions (Ca^{2+})

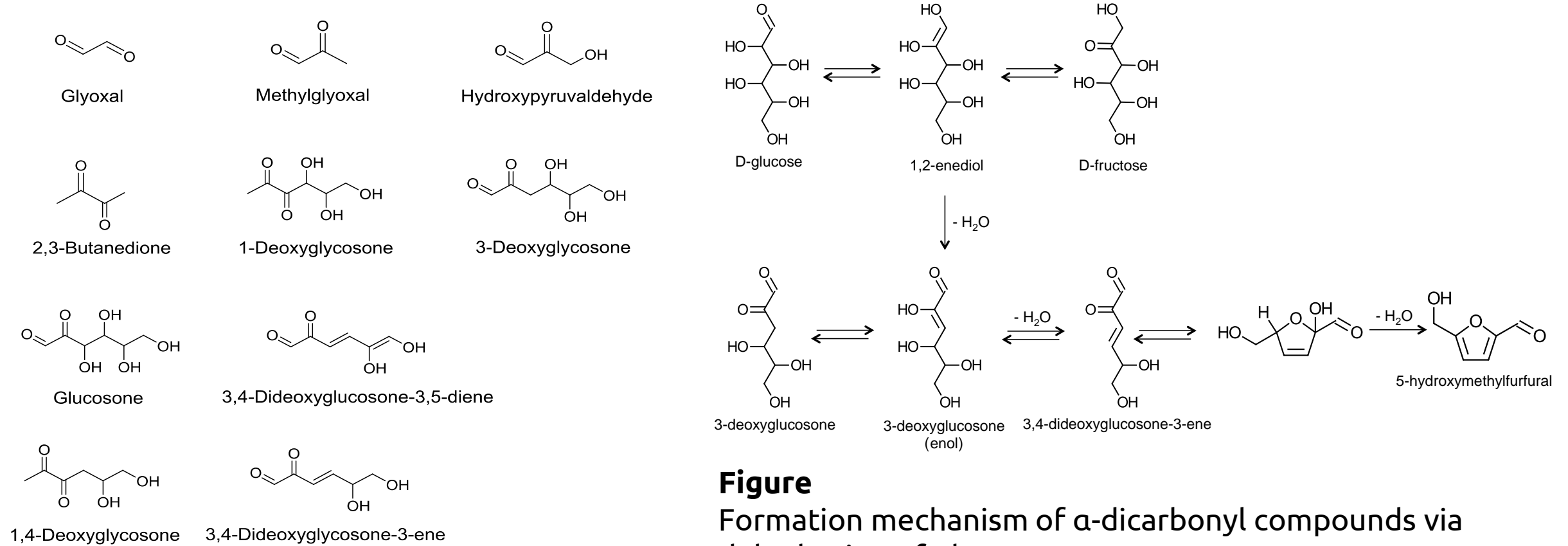
Dicarbonyl Compounds

Maximum limits are not regulated in EU.

Dicarbonyl compounds form advanced glycation end (AGEs) products that play a role in diabetes, aging and Alzheimer's diseases.



Dicarbonyl Compounds



Figure

Formation mechanism of α -dicarbonyl compounds via dehydration of glucose

Figure

Examples of the structures of α -dicarbonyl compounds generated in foods because of Maillard reaction, sugar decomposition and lipid oxidation reactions

Hidroksimetilfurfural

Maximum limits are regulated in EU.

- 40 mg/ kg in general
- 80 mg/kg in Tropical honey

Sugar Rich Products

HMF is formed in the Maillard reaction as well as during caramelization. HMF can form in sugar-containing food, particularly as a result of heating or cooking. It is also slowly generated during storage. Acid conditions favor generation of HMF. Its formation has been the topic of significant study as HMF was regarded as being potentially carcinogenic to humans.

HMF

Sugar
degradation
pathway
↓

Maillard
reaction
pathway
↓

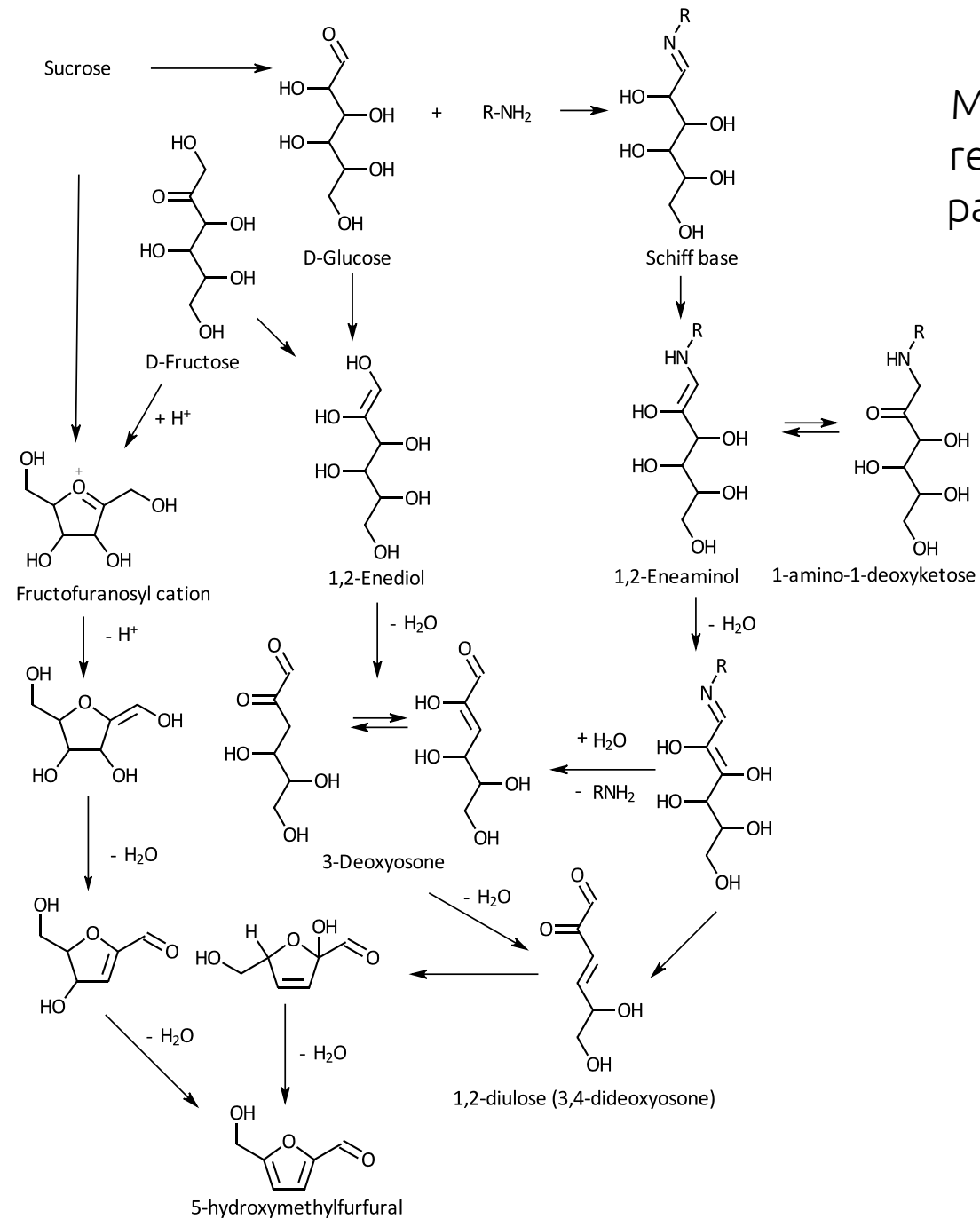


Figure
Formation mechanism of HMF

Heterocyclic Amines

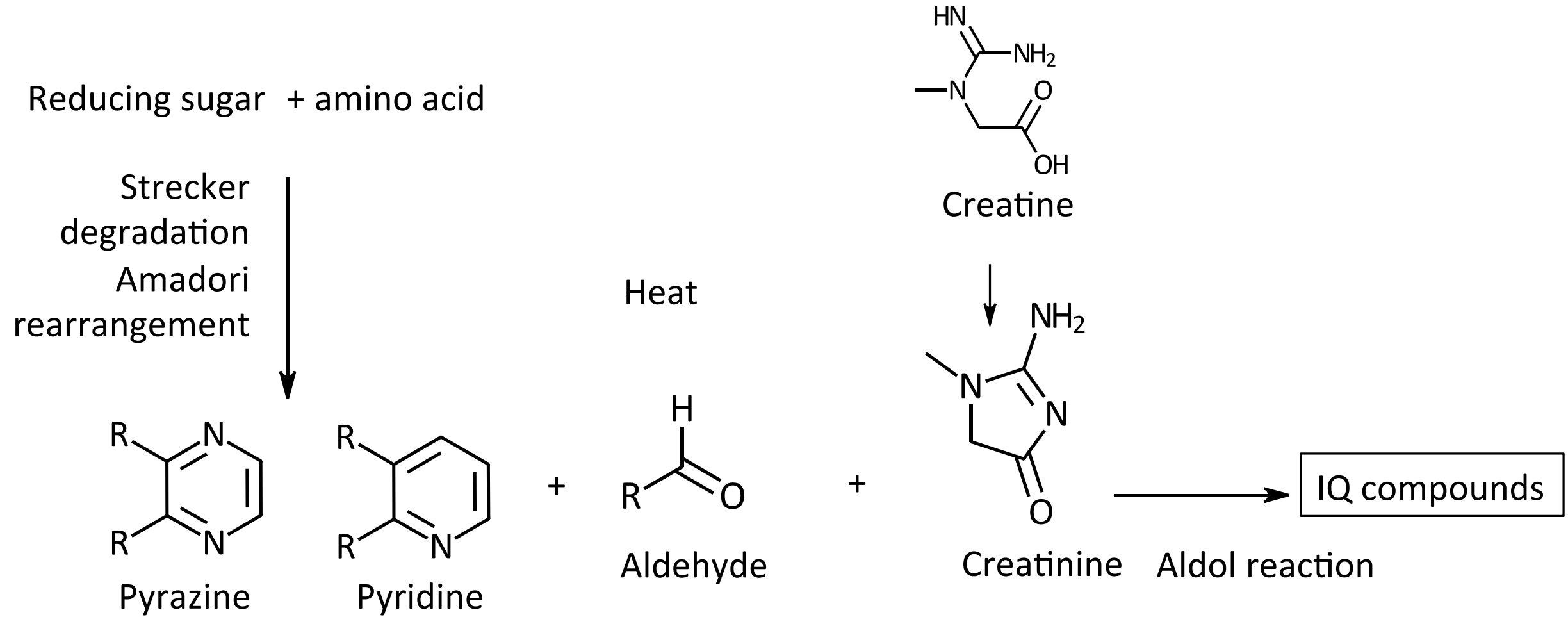
Maximum limits are not regulated in EU.

Roasted Meats

Heterocyclic amines (HCAs) are a group of mutagenic compounds found in cooked meats, particularly well-done meats.

HCAs in the form of **amino carboline** or **amino imidazo-azaarene** groups are formed during frying, broiling, smoking, or barbecuing meat and fish at high temperatures due to the abundance of amino acids or creatine in these foods.

Heterocyclic Amines



Figure

Formation mechanism of heterocyclic amines in foods via the Maillard reaction

Take home message

Eating is not free of risk

A modern laboratory setting with multiple workstations. Each workstation features a computer monitor displaying data graphs, a keyboard, and a mouse. The lab is equipped with various scientific instruments, including a large stack of white machines on the left and robotic arms with glass vessels suspended from the ceiling. Large windows on the left side provide natural light and a view of trees outside. The floor is made of light-colored tiles, and the overall environment is clean and professional.

Food Quality & Safety